

# Applying molecular microbiological methods to understand and prevent microbiologically influenced corrosion

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# How Bacteria Cause Corrosion?

- Varies significantly from metal to metal & type of bacteria
  - Slime producing microbes cause differential aeration cells / under deposit corrosion
    - Fungi, Algae, Protozoa, Bacteria
    - Provide microclimate for sulfate reducing bacteria
  - Aerobic bacteria may oxidize iron
  - Aerobic acid producing bacteria produce organic acids
  - Sulfate reducing bacteria produce  $H_2S$  and/or sulfuric acid
  - Depolarization of the cathode by hydrogen-utilizing bacteria

# Consequences of Bacteria in Shale Production

- Corrosion of pipelines
- Corrosion of facilities
- Souring of reservoirs
- Plugging of filters

# Corrosion Rates from MIC

## RO Water Piping

- 6" Sch 10S 304/304L Line (0.148")
- Hydrotest water: potable water stored in an open air tank
- Water filled: 3 weeks
- Operated: 3 weeks
- Corrosion Rate:
  - Assume 6 weeks: 1286 mpy

## Utility water Piping

- 4" 304L stainless steel, 0.213" thick
- Leaked after several months
- Hydrotest water: chlorinated potable water
- Corrosion Rates:
  - Assume 3 months: 852 mpy
  - Assume 6 months: 426 mpy

## Pipeline

- 18" carbon steel seamless pipeline, 0.312" thick
- Hydrotest water: Untreated creek water
- Water filled: several months
- Drying: via gel pig
- Sat empty: 2 months
- In service: 3 years
- Corrosion Rates:
  - Assume 3.5 years: 90 mpy
  - Assume 5 years: 62 mpy

Take away: Stainless Steels are much more rapidly affected than carbon steel

# Primary Standard Associated with Bacterial Testing

NACE TM 0194 -2014 - Field Monitoring of Bacterial Growth in Oil and Gas Systems

- Sampling Procedures for Planktonic Bacteria
- Assessment and Sampling of Sessile Bacteria
- Culture Techniques
- Evaluation of Chemicals for Control of Bacteria
- Non-Media-Based Field Methods
- Select Appendixes
  - Alternative Methods for Assessing Bacterial Populations
  - Growth Media Formulations

# Basics of Bacteria Detection

## Media Based

- Serial Dilution – MIC kit tests

## Non-Media Based

- ATP - Adenosine triphosphate
- APS - Adenosine 5 Phosphosulfate reductase
- Hydrogenase Measurement

## Molecular Microbiological Methods (MMM)

- FISH - fluorescent *in situ* hybridization
- qPCR – Quantitative real-time polymerase chain reaction
- DNA Sequencing by PCR-DGGE or TGGE - Polymerase chain reaction denaturing gradient gel electrophoresis or thermal gradient gel electrophoresis
- Next Generation Sequencing (NGS)

# Serial Dilution

- Need living bacteria
- 30 day wait for results
- Everyone in oil field knows what serial dilution is except the new people
  - API Bug Bottles
    - RP 38 – Biological Analysis of Subsurface Injection Waters
    - Written 1975
    - Withdrawn 1982?
  - NACE
    - TM 0194 -2014 Field Monitoring of Bacterial Growth in Oil and Gas Systems
    - TM0106-2006 Detection, Testing, and Evaluation of Microbiologically Influenced Corrosion (MIC) on External Surfaces of Buried Pipelines

# Serial Dilution

1 ml of fluid to be sampled

Remove 1 ml of fluid & Inject

Remove 1 ml of fluid & Inject

Remove 1 ml of fluid & Inject

Remove 1 ml of fluid & Inject

Remove 1 ml of fluid & Inject



Growth Media will change color to signal bacteria viability



# Serial Dilution

Sulfate Reducing Bacteria



Small piece of steel in SRB bottles which reacts with  $H_2S$  to form  $FeS$

Acid Producing Bacteria



Phenol red changes to orange or yellow depending on the pH

GHB bottles will turn cloudy if bacteria are present

# Culturing

- Salinity should match process
- Temperature should match process (+/- 5 C)
- Bottles should be inoculated on site / in the field
- If not inoculated on site, sample should be refrigerated if more than 48 hours to transport from the field
- SRB incubate for 28 days
- Other bacteria incubate for 14 days
- For SRB testing, should a vial turn black within 2 hours of sampling, it is assumed that sulfides in the sample caused the color change

# Serial Dilution – Simplistic Estimate of Population

# of Bottles Turned	Range of Viable Bacteria per mL of Sample
1	1 to 10
2	10 to 100
3	100 to 1000
4	1,000 to 10,000
5	10,000 to 100,000
6	100,000 to 1,000,000

# Serial Dilution – Statistical Estimates

# of Bottles Turned	Dilution of Sample	Estimated Range of Bacteria per mL	# of Bottles Turned	Dilution of Sample	Estimated Range of Bacteria per mL	# of Bottles Turned	Dilution of Sample	Estimated Range of Bacteria per mL
Single Sample			Duplicate Samples			Five Replicate Samples		
1	1:10	1 to 145	1	1:10	1 to 66	1	1:10	1 to 33
2	1:100	7 to 1,450	2	1:100	15 to 660	2	1:100	33 to 330
3	1:1000	69 to 14,500	3	1:1000	150 to 6,600	3	1:1000	330 to 3,300
4	1:10,000	690 to 145,000	4	1:10,000	1,500 to 6,6000	4	1:10,000	3,300 to 33,000
5	1:100,000	6,900 to 1,450,000	5	1:100,000	15,000 to 660,000	5	1:100,000	33,000 to 330,000
6	1:1,000,000	69,000 to 145,000,000	6	1:1,000,000	150,000 to 6,600,000	6	1:1,000,000	330,000 to 3,300,000

# Problems with Serial Dilution

- Contamination
- Samples degrade over time
- Takes 28 days for reliable SRB results
- Culturing should occur at temperature of process
- Culturing should occur at salinity of process
  
- “While great advances have been made in cultivation techniques, it is now accepted that only 0.001-15% of the viable bacteria are culturable by these classical microbiological methods<sup>1</sup>.”
  - Corrosion 08652 *Molecular Identification of MIC Bacteria from Scale and Produced Water: Similarities and Differences*, Jan Larsen et al

# Non-Media Based Field Methods

- General Population
  - ATP Photometry
- Sulfate Reducing
  - APS Reductase Measurement
  - Hydrogenase Measurement

# ATP - Adenosine triphosphate

- “ATP is a molecule found in and around living cells, and as such it gives a direct measure of biological concentration and health. ATP is quantified by measuring the light produced through its reaction with the naturally occurring firefly enzyme luciferase using a luminometer. The amount of light produced is directly proportional to the amount of ATP present in the sample.”
  - [https://en.wikipedia.org/wiki/ATP\\_test](https://en.wikipedia.org/wiki/ATP_test)

# ATP Photometry

- Produced by all living cells – degrades quickly upon death
- Can be measured with light
- Does not distinguish between types of bacteria
- Does not distinguish between other living cells
- No direct conversion of ATP concentrations to microbial concentrations\*
- Used for trending, measuring biocide efficacy



# APS - Adenosine 5 Phosphosulfate Reductase Measurement

- SRB's possess a unique enzyme, APS - Adenosine 5 Phosphosulfate reductase measurement
- No growth media
- Independent of salinity, temperature, and redox condition
- Color change proportional to concentration of APS-reductase present
- Testing takes minutes
- Brand names: Rapid Chek, Quick Chek

# Hydrogenase Measurement

- Bacteria which use hydrogen as part of their metabolism produce a hydrogenase enzyme
- Quantifies the activity of the hydrogen utilizing bacteria
- No growth media
- Performed on filters or sessile samples
- Independent of salinity, temperature, and redox condition
- Color change proportional to concentration of hydrogenase present
- Testing takes hours
- Testing is non-specific – will reflect the general population active hydrogen utilizing bacteria, not specific species

# Quantitative Fluorescent *In Situ* Hybridization (qFISH)

- Stain cells of interest with fluorescent dye
  - Stains may target specific groups
- Perform epifluorescence microscopy at 1000X
- Count the bacteria
- Detects active bacteria, not inactive or dead
  - DAPI – a generic stain that stains all organisms present (bacteria & Archaea)

# Quantitative Polymérase Chain Reaction (qPCR)

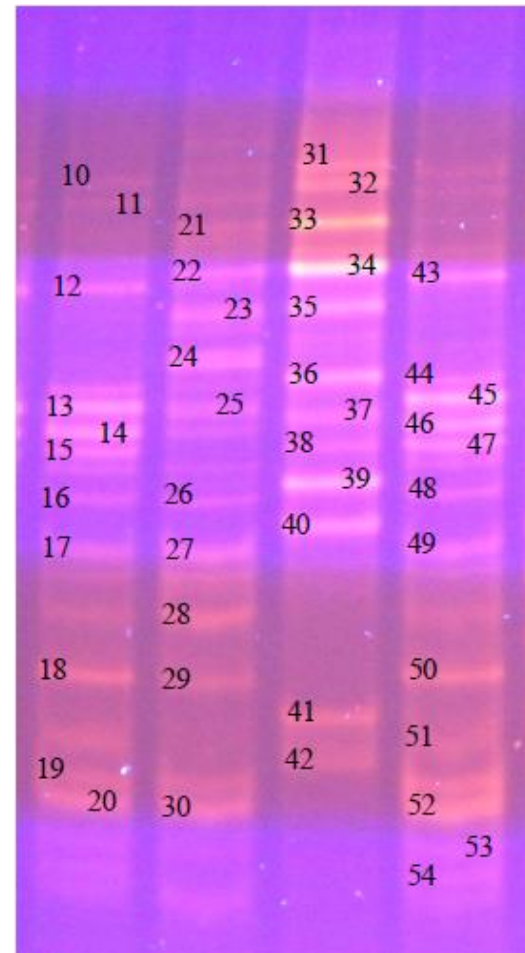
- The polymerase chain reaction (PCR) is a technique used in molecular biology to amplify a single copy or a few copies of a piece of DNA across several orders of magnitude, generating thousands to millions of copies of a particular DNA sequence.
  - [https://en.wikipedia.org/wiki/Polymerase\\_chain\\_reaction](https://en.wikipedia.org/wiki/Polymerase_chain_reaction)

# Quantitative Polymerase Chain Reaction (qPCR)

- Amplifies specific gene sequences from target organisms
- Genetic material is extracted from a sample
- Quantify the copies of a specific gene
- Does not distinguish between live, inactive or dead cells
- Quantify the total amount of bacteria and subspecies against a known sample
- Cost \$350 / sample at Company A + disposal + reporting
  - Total Bacteria, SRB, APB, Archaea
- Cost \$900 / sample at Company B
  - SRB, APB, Methanogens, IRB, Metal Oxidizing, SOB, Denitrification, Nitrification, Nitrogen Fixing, Slime Producing, Total Bacteria

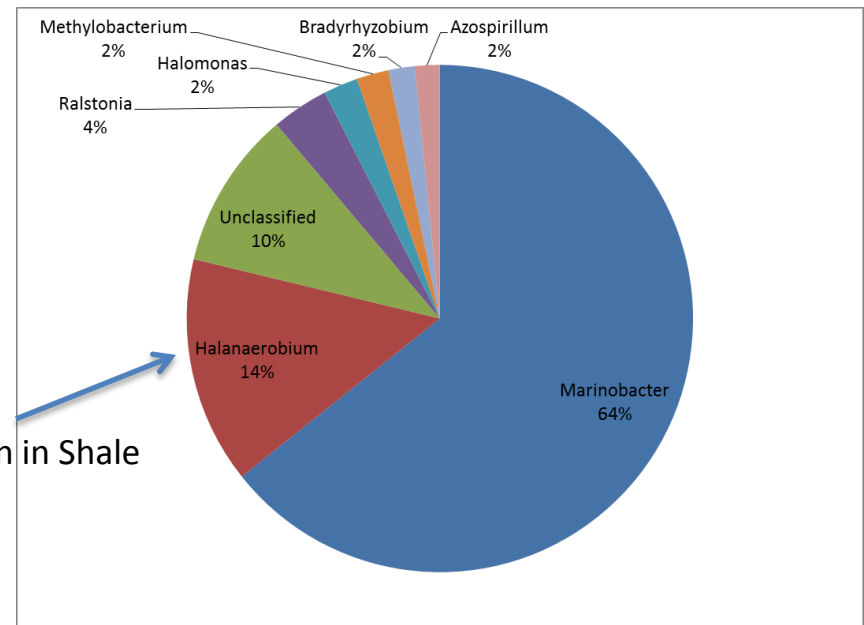
# PCR-DGGE - Polymerase chain reaction denaturing gradient gel electrophoresis

- PCR-DGGE - PCR is used to replicate a gene fragment (usually) 16S rRNA
- Denaturing gel (DGGE) or temperature (TGGE) causes gene fragments to spread, creating a unique fingerprint
- [https://en.wikipedia.org/wiki/Community\\_Fingerprinting](https://en.wikipedia.org/wiki/Community_Fingerprinting)



# DNA Sequencing

- “DNA sequencing of bacterial 16S rRNA gene fragments”
  - Fingerprint generated via PCR-DGGE or PCR-TGGE
  - Compared to a standard database to identify specific genus and species of bacteria present
- Cost – ~\$4000, or \$1000/sample with a minimum of 5 samples
- Good for determining effect of biocide treatments
  - Halanaerobium – resistant to gluteraldehyde



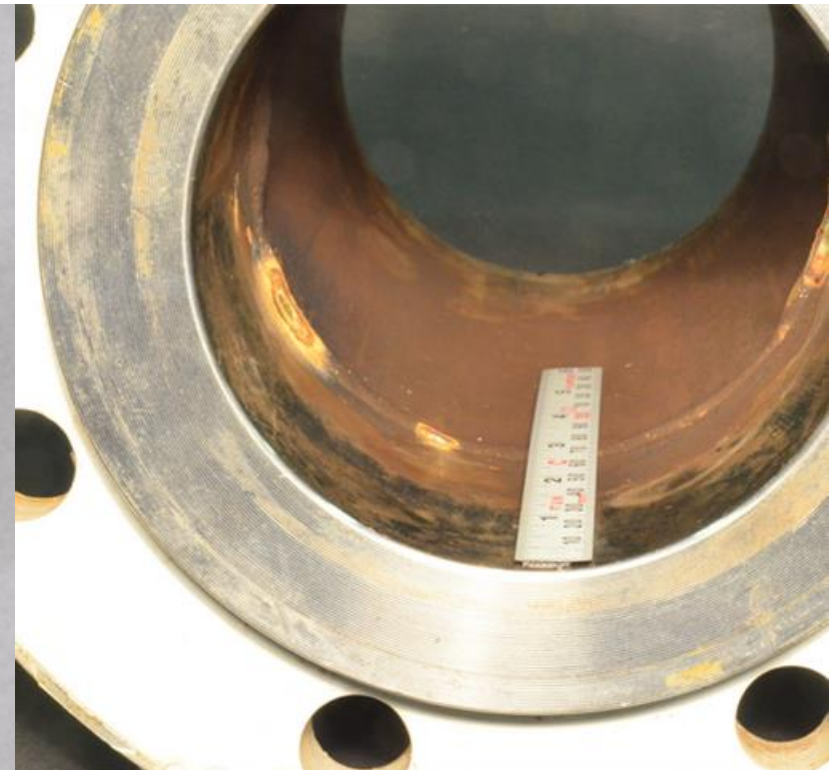
# Field Failure Example #1

- Land based facility in Central US
- Water comes from
  - City potable water
  - Well A
  - Well B – Known to have oil contamination
  - Blended to keep chloride concentration around 200 ppm
- 316 Piping Failed
  - New water plant, started up in July 2015 operates at 60 F
  - 20+ Leaks
  - April 2016 sample sent for failure analysis
  - Hypochlorite dosage 1 ppm target
    - Known to be as low as 0.5 ppm or as high as 4 ppm



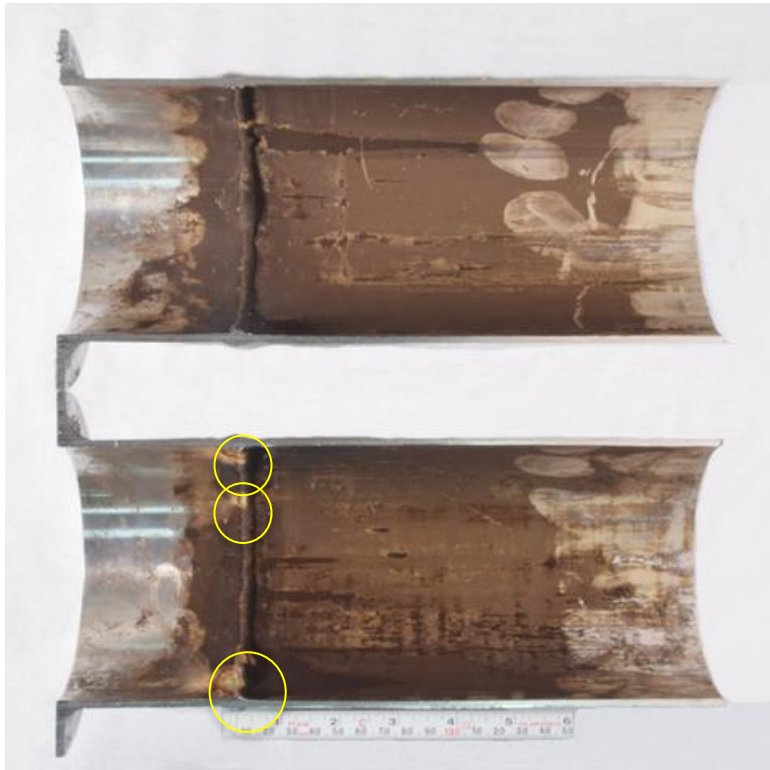
# Field Failure Example #1

Sample Received at SES



# Field Failure Example #1

Interior View



Three pits identified

Exterior View

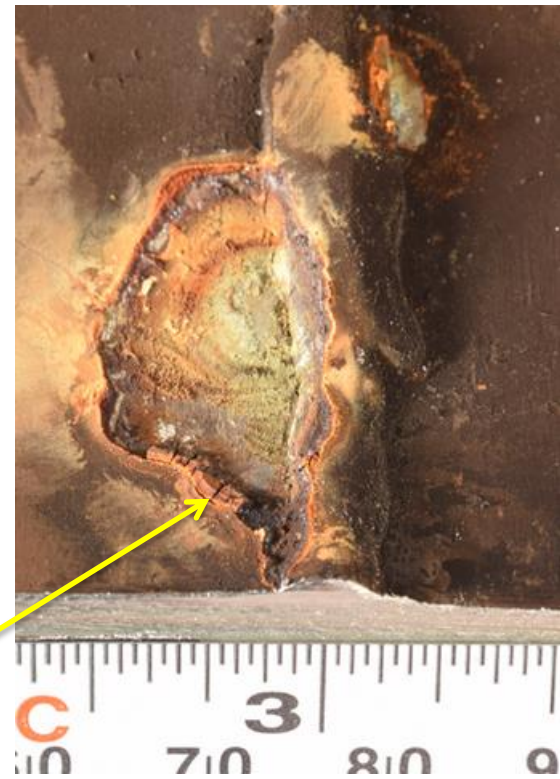


Significant Iron Contamination

# Field Failure Example #1



Classic Tubercle Crust



# Field Failure Example #1

Metallographic Sample Through Pit #1



# Field Failure Example #1

## qPCR Results

Sample	Total Bacteria	SRB	APB	Archaea
Non-corroded area	$3.7 \times 10^4$	$5.7 \times 10^2$	$2.0 \times 10^1$	$< 10^1$
Undercut pit sample	$3.1 \times 10^5$	$7.5 \times 10^3$	$1.1 \times 10^2$	$< 10^1$

Hypo-chlorite is not enough

# Field Failure Example #1

## DNA Results

Genus	% Abundance	Species Detected	
Marinobacter	54.2	sediminum, santoriensis, aquaeoli	Iron Oxidizing
Halanaerobium	12.2	saccharolyticum, praevalens	Anaerobic (no oxygen)
Unclassified	8.5		
Ralstonia	3.06	insidiosa	
Halomonas	1.86	fontilapidosi	
Methylobacterium	1.73		
Bradyrhizobium	1.41		
Azospirillum	1.34		

# Comments on Example #1

- Reputable firm designed and built the facility
- Hypo-chloride rates, salinity and temperatures matched published laboratory data (metal manufacturer) aimed at justifying the use of 316 under these conditions
  - Samples in study were pickled and passivated
- Heat tint, iron contamination and bacteria all contributed to the problem

# Prevention

- Recognize the risk
  - Check for a written water standard
    - Does it address MIC, souring?
  - Plan for chemical treatment
  - Understand the benefits of filtration
- Understand source water & your chemical treatment
  - Focus is often on chloride content, not bacteria
  - Potable water is clean, but may need treatment
  - Pond water, bay water & seawater are generally not clean
    - Definitely need treatment
  - Understand if biocide is affected by oxygen
    - THPS, Glutaraldehyde will react with oxygen
  - Disposal of treated water / exposure of personnel



# Open Discussion

## Questions, comments?

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